

6th International Conference on Physical and Theoretical Chemistry

8th World Congress on Bio-Polymers and Polymer Chemistry

February 22, 2022 | Webinar

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<u>Novel catalytic transformations of ionic polymer-supported</u> palladium metal nanoparticle compounds

etal nanoparticles are increasingly used in catalysis due to their high surface area and increased density of surface active sites compared to bulk metal.1 Ionic liquids have been shown to stabilized nanoparticles towards aggregation and also, in some cases, to enhance the catalytic activity or selectivity.2 In this project cross-linked ionic copolymers were prepared by radical polymerisation of imidazoliumfunctionalised styrene monomers together with styrene carrying a neutral pyrrolidinone derivative designed to interact with metal nanoparticles. The ionic polymer was loaded with PdCl42- by anion exchange and then hydrogenation produced a Pd (0) loaded polymer which was characterized by TEM, microanalysis, ICP-OES, XPS, XRD, and SEM. These palladiumimmobilised ionic polymer-supported nanoparticles were demonstrated to be active catalysts for Suzuki cross coupling reactions, and we will explore the efficiency of palladium nanoparticles in a broader range of reactions including the selective hydrogenation of α , β -unsaturated aldehydes and ketones as well as the decompositions of formic acid to CO2 and H2.



 $\label{eq:ar-Br} \begin{array}{c} \operatorname{Ar-Br} & \leftarrow & \overbrace{}^{B(OH)_2} & \xrightarrow{\operatorname{cat.} (0.1 \ \operatorname{moP}_{4})} \\ & & \overbrace{}^{K_2CO_3} \\ & & \overbrace{}^{EtOH-H_2O, \ 30 \ \circ C} \end{array} \xrightarrow{} \qquad \operatorname{Ar} \end{array}$

Scheme 1: The general conditions for Suzuki coupling.

Speaker Biography

Hind A has completed her PhD in 2019 from Newcastle University, UK. She is the Assistance professor at Newcastle University, SA. She has over 12 publications that have been cited over 46 times.

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Figure1: The structure and SEM image of the palladium loaded ionic polymer.

Received date: April 4, 2022; Accepted date: April 6, 2022; Published date: May 30, 2022